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Dry Ice Lab Conclusion

Dry ice is the solid form of Carbon Dioxide and it is extremely cold which makes it dangerous to handle it without any protection. In this lab, a small amount of dry ice was placed in a pipette bulb and held tightly shut with a pair of tweezers. The pipette bulb was placed in a cup filled with room temperature water. After a few seconds, a small explosion occurred. Inside the pipette bulb, the dry ice was in a solid and a gas state. As the particles of the gas expanded and couldn’t be contained in the pipette bulb this caused a little explosion where the gas escaped.